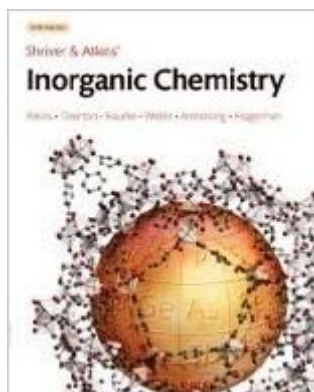


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Inorganic Chemistry 5th (fifth) Edition



Synopsis

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Customer Reviews

Very clean and VERY NEW!!!! fast shipping guaranteed!!

I have to mention that that the author, Atkins, also writes textbooks for physical chemistry. That partially explains why the content is taught in a linear fashion with mathematical formulas. It seems to give only brief explanations of certain topics, where the assigned problems, in the book, ask questions where you are required to use equations not even mentioned in the textbook. This textbook feels incomplete; there isn't even a mention of Slater's rule or using Vsepr theory alongside MO theory. It is definitely lacking.

The explanations are pretty good. However they are a bit lacking for some fundamental and difficult concepts such as constructing MO theory orbitals and how it is done. The first chapter is rather dry and easy to just skip, but should just be review from lower level chemistry courses. I will try to update this review as I follow the book through the class. Thankfully my professor is amazing. Edit: Further in the semester and decreasing the score from a 4 to a 3. The book barely touches on hard/soft acid/base chemistry which is a huge topic for predicting reactions. Some theoretical topics are covered in great detail, but are wholly useless for the level this book is written at. They explain lattices to great detail but lack sufficient explanation for all the ways Born-Haber cycles can be used. Equations aren't always presented well (not in their own paragraph) which makes reading difficult. Example problems aren't well explained. Basically, the book covers information very well, but it doesn't cover the information it needs to cover for a course well enough.

As a chemistry major, I hate this book. WAY too much jargon, extremely difficult to understand, very long passages in almost fine print (hurt my eyes). Very few figures to really explain chemistry (most

chemistry is visual) and this book does a horrible job explaining what you are looking at, sometimes it doesn't even explain it. Better off using google or a different book. At least the book comes in great condition. (because why even open this piece of garbage)

Great book, (I also found the pdf of it online). It really covers everything very nicely and it's very clearly laid out. I like how the layout helps break up the pages so that it's not like just looking at a block of dry text (inorganic and physical chem books tend to do that a lot - See Ira Levine, Physical Chemistry 6th).

The biggest problem with the paperback version of this textbook is that it is printed in black and white but the figures use color to differentiate between items. Not being able to figure out which line is supposed to be red or which is supposed to be blue, for example, makes understanding these already dense discussions even more difficult.

Was an indication that a new article, not visible only to the used book. There are discolored and according as open many times. This is not New.

As my title describes, I am completely dissatisfied with this book. It is vaguely written. It does not clearly and concisely explain concepts or questions. It is written as if it is NOT an introductory textbook. It assumes you know principles that an introductory student does not know, which contributes to the poor explanations. The practice questions in the book are vague as well. Most of the time, it is difficult to understand what is even being asked and the wording is confusing rather than clear and to the point. There are multiple mistakes in the book and answer key (the answer key has most of the mistakes and explains things with concepts not even described in the book, but I'll save its comments for its own review). I would not recommend this book. I would actually strongly discourage it and recommend "Inorganic Chemistry" by Gary Miessler or Catherine Housecroft.

Ordered book that was listed to be in good condition. When the book arrived it was far from good condition. The spine is badly torn and the beginning pages are about to fall out because of this. The actual condition of the book was not reflected in the description purely to charge more for the book.

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